Name: D	Date:
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Mixtures (U.10b)

Vocab matching: write the letter of the definition to the left of the correct vocabulary term.

	1. mixture	a. Medium-sized solute that will NOT separate into layers under force of gravity
	2. heterogeneous	b. A mixture that has the same consistency throughout
	3. suspension	c. the solute is visible to the eye and will separate into layers under force of gravity
	4. solvent	d. Two or more things physically combined
	5. homogeneous	e. The part of a solution that is being dissolved and is in the smaller proportion
	6. alloy	f. A dissolvable substance
	7. colloid	g. A mixture that does NOT have the same consistency throughout
	8. solution	h. The part of a solution that does the dissolving and is in the largest proportion
	9. solute	i. A solid+solid solution such as brass or steel
_	10. insoluble	j. The smallest sized solute particles (even lasers cannot image them for us); they will NOT separate into layers under force of gravity
	11. soluble	k. The universal solvent
	12. Pure substance	I. A substance that will not dissolve
	13. Water (H ₂ O)	m. an element or compound

Classify the following substances as solutions, colloids, or suspensions:

14	Salad dressing	
15	Coca-cola	
16	Fertilizer spread on a lawn	
17	House paint for walls	
18	A garden compost pile	
19	A ring that is 20% gold and 80% copper	
20	Laundry detergent (contains white and blue crystals)	
21	Cough syrup that must be shaken before pouring	
22	Fresh-squeezed orange juice	
23	Benzene (C ₆ H ₆)	
24	Nail polish remover (C ₃ H ₆ O)	
25	A muddy pond	
26	milk	
27	Fresh-brewed coffee	
28	Wood smoke	

Answer the following questions to the best of your ability; use your notes and Google searches to improve your answer.

29.	Not all solutions	are solids dissolve	d in liquids.	Provide two	examples of	of other ty	pes of
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- 30. What is the Tyndall Effect (look it up!)? Give a common example of this effect.
- 31. What kind of mixture is represented by our skin moisturizer prototypes? What evidence supports your claims?